



Transformation to the production of REE oxides through hydrometallurgical routes

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Géosciences pour une Terre durable



AGENDA

- 1. Context
- 2. The process
- 3. Characterization of the HDD NdFeB magnets
- 4. Leaching of NdFeB powder using organic acids
- 5. Selective leaching using organic acids
- 6. Conclusions and perspectives







CONTEXT

Rare earth permanent magnets:

• One of the most important applications of REEs

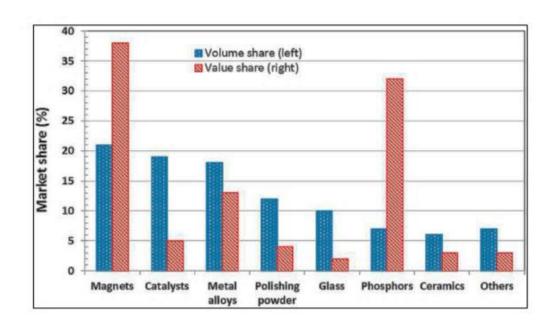
Neodymium magnets (NdFeB) :

- Has the highest energy density
- Used in several high-tech products, wind power turbines and electric vehicles
- Contains: Nd, Pr, Dy, Tb, Gd

Recycling potential of NdFeB magnets:

- Contains metals identified as critical by the European Commission (2011): Nd, Dy, Co, Tb
- Importance of current stocks of spent NdFeB magnets
- Continuous increase in stocks due to growth in the wind power and electric transportation sectors









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Market share of REEs applications.

CONTEXT

- Direct reuse
- Hydrogen decrepitation
- Hydrometallurgy
- Pyrometallurgy
- Resintering...

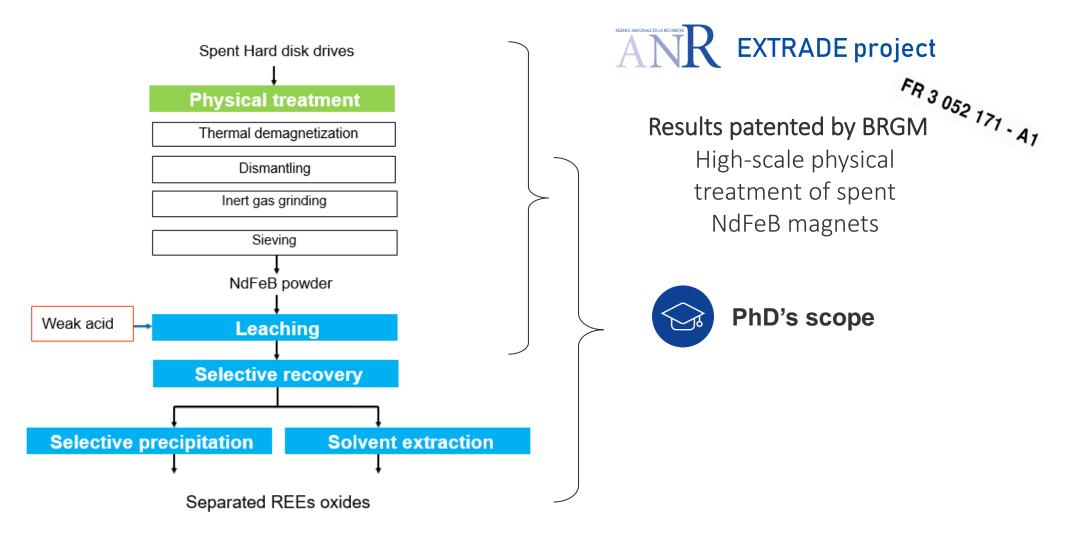
- Applicable to all types of waste.
- Can handle large variations in composition/impurities
- Similar to virgin mining of REEs







THE PROCESS





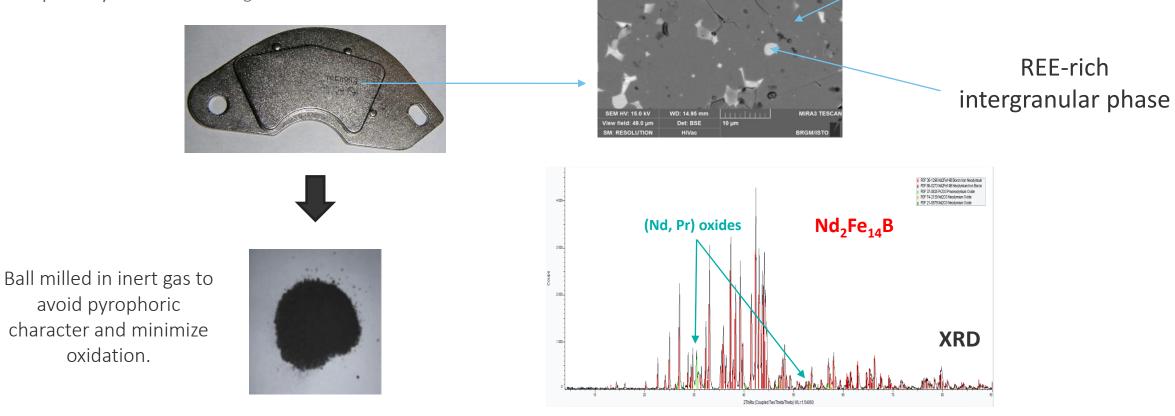


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CHARACTERIZATION OF THE HDD NDFEB MAGNETS

Sample study: HDD's NdFeB magnet.



SEM/ED

Elei	ment	В	Со	Dy	Fe	Nd	Ni	Pr
%	ő wt	0.9	1.5	1.2	62.6	22.8	0.6	3.3







Nd₂Fe₁₄B

LEACHING OF NDFEB POWDER USING ORGANIC ACIDS

In a hydrometallurgical process, leaching is the process of dissolving certain metals from the ore or waste.

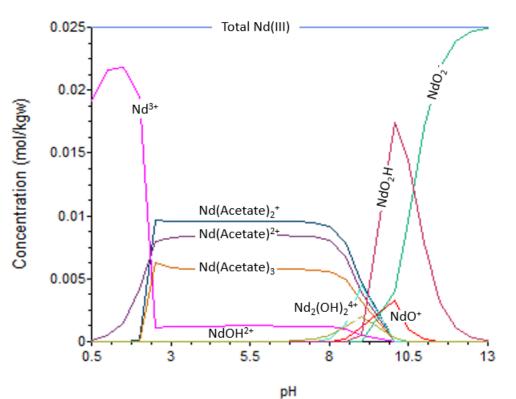
Mineral acid, such as: H₂SO₄, HCl, HNO₃

 $TR_{(s)} + 6H^{+}_{(aq)} \leftrightarrows TR^{3+}_{(aq)} + 6H_{2(g)}$ $TR_{2}O_{3(s)} + 6H^{+}_{(aq)} \leftrightarrows 2TR^{3+}_{(aq)} + 3H_{2}O_{(g)}$ $TR(OH)_{3(s)} + 3H^{+}_{(aq)} \leftrightarrows TR^{3+}_{(aq)} + 3H_{2}O_{(l)}$

Eco-friendly alternatives: organic weak acids

Acetic acid : $Nd^{3+} + jCH_3COO^- \Leftrightarrow Nd (CH_3COO)_j^{(3-j)+}, j = 1, 2, 3$ Citric acid : $nNd^{3+} + jH^+ + kCit^{3-} \Leftrightarrow Nd_kH_jCit_k^{(3n+j-3k)}$ Presence of NdCit, NdHCit, NdHCit₂, NdCit₂ in a pH range of [2-5].

Easier to handle, form less toxic gases, easier biodegradability, possible production from agro-industrial waste...



Aqueous speciation of Nd(III) as a function of pH in diluted acetic acid using pHreeqC (Database: LLNL).







LEACHING OF NDFEB POWDER USING ORGANIC ACIDS

72.90

63.00

53.10

43.20

33.30

23.40

13.50

3.600

35.12

30.35

25.58

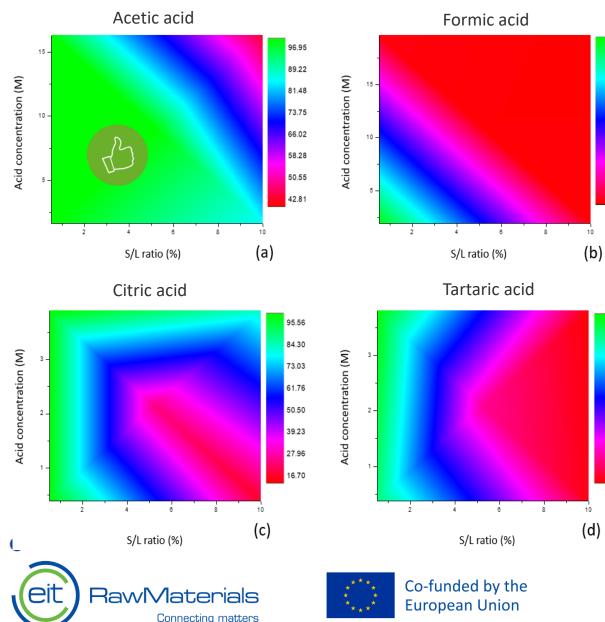
20.81

16.04

11.27

6.504

1.734



Nd, Pr and Dy have the same leaching behavior in all tested acids

Acetic acid: Best candidate for leaching REEs under industrially favorable conditions; high S/L ratios and low acid concentrations

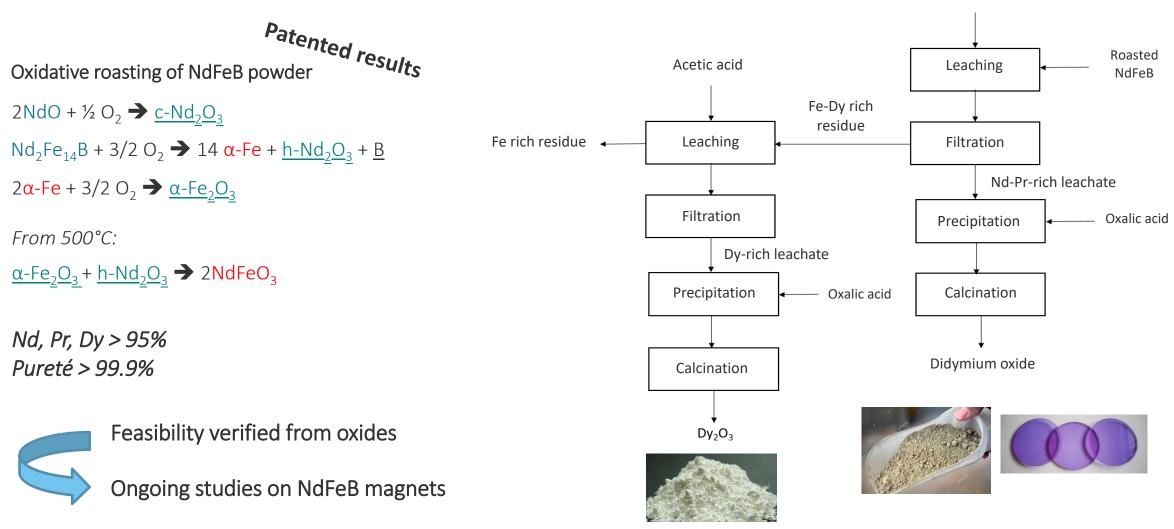
> 90% of REEs leached:S/L ratio (%) [0.5 - 5]Acetic acid concentration (M) [1.6-10]

Partial/ total co-leaching of Fe, Co and B

Formic acid: Precipitation of REEs in formates Tartaric acid: Precipitation of REEs in hydroxides



SELECTIVE LEACHING USING ORGANIC ACIDS





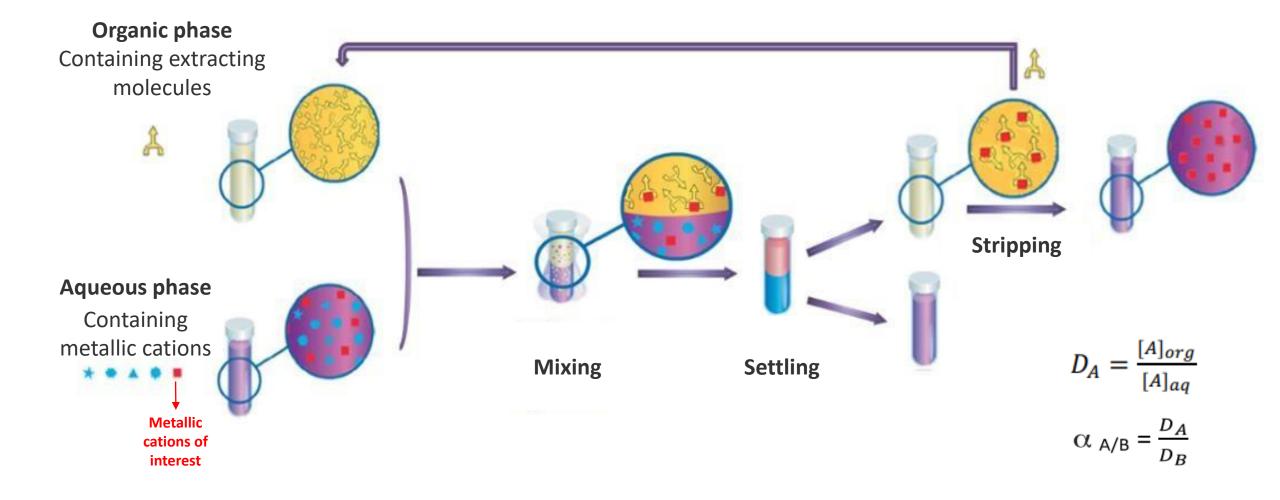


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Acetic acid

SOLVENT EXTRACTION ON ORGANIC ACID LEACHATES







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CONCLUSIONS AND PERSPECTIVES

Characterization

- Microstructure: Matrix phase (Nd₂Fe₁₄B), intergranular phase (REEs oxides)
- Chemical composition: 63% Fe, 23% Nd, 1% Dy, 3% Pr, 1% B

Leaching

- Acetic acid: Efficient weak acid to leach REEs in favorable conditions
- Feasible intra-REEs selective leaching on oxidized NdFeB powder







Thank you for your attention

Questions?





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